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## The Periodic Table Review Worksheet

How many grams of C are in 3.2 moles of C?

How many moles of S are in 4.5 grams of S?

If you had 8.1 grams of magnesium how many moles of magnesium would that be?

If you had 9.6 moles of bromine how many grams of bromine would that be?

How many molecules of F are in 3.7 grams of F?

How many moles of Li are contained in 3.6 \*  $10^{25}$  molecules of Li?

If you had 12 atoms of hydrogen how many grams of CH<sub>4</sub> could you make?

If you have 56 grams of  $Na_2I_3$  then how many molecules of  $Na_2I_3$  is that?

With 4.6 \*  $10^{-23}$  moles of Pb<sub>2</sub>N<sub>3</sub> how many molecules of Pb<sub>2</sub>N<sub>3</sub> will you have?

With 7.9 \*  $10^{13}$  grams of Ca<sub>3</sub>B<sub>5</sub> how many atoms of boron will you have?

If you have 54 atoms of Be how many moles of BeO can you make?